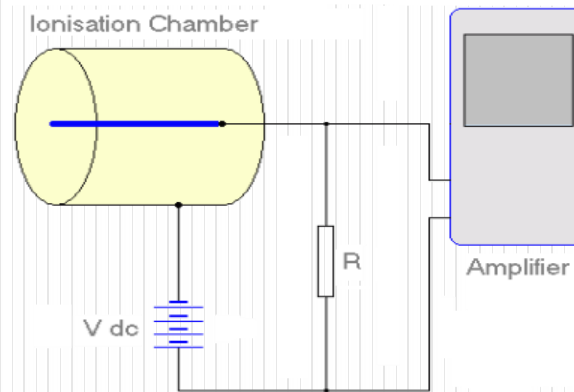


RADIATION DOSIMETRY

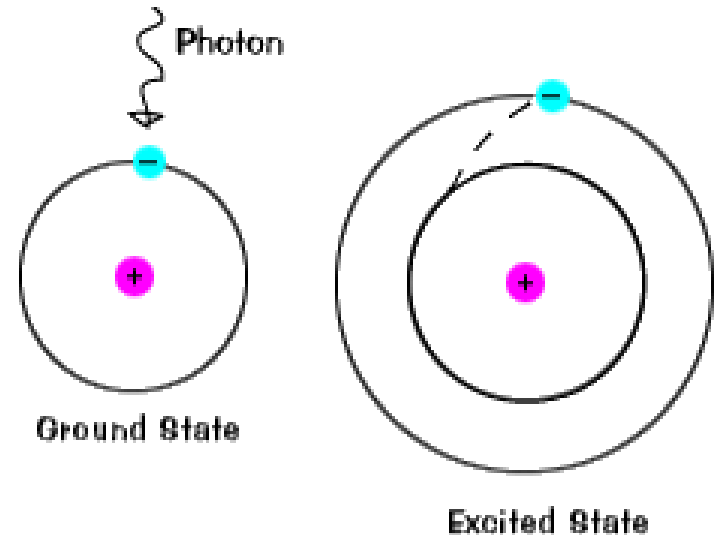
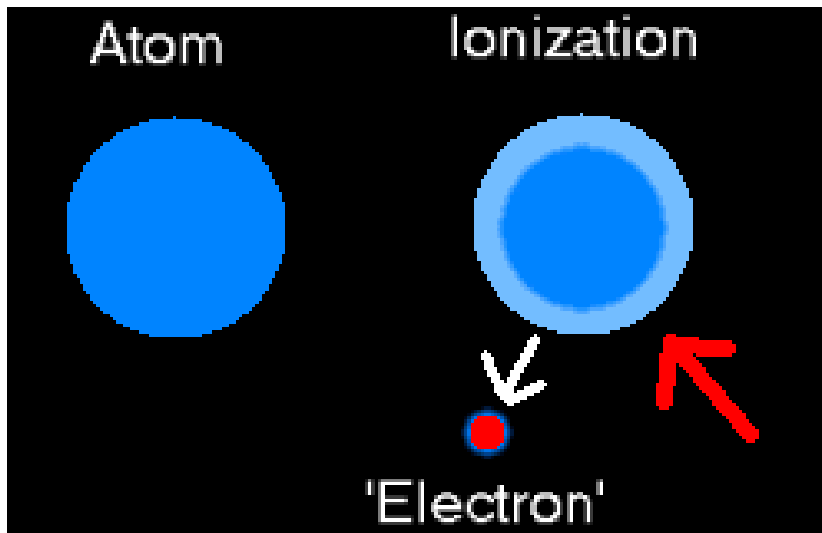


By

Dr. Ahmad Mokhtar Abodahab

Absorbed dose

- Effects of ionizing radiations → correlated with the energy deposited as **ionization** and **excitation** of atoms of the material.



- **Absorbed dose:** energy deposited per unit mass of the material (in **joules / Kg**).

SI unit of Absorbed Dose

Gray (Gy)

- $1(\text{Gy}) = \text{J} / \text{kg}$
- **Absorbed dose rate** \rightarrow is measured in grays / second,
- with the usual **multiples** and **sub-multiples**.
- **Absorbed dose** : is used to define the quantity of radiation delivered at a specified point in radiation field.
- It is valuable when considering tissue and biological effects.

SI unit = International System Units

(Joule الجول)

الوحدة الرياضية/الفيزيائية لقياس الطاقة

- **Before 1980,**

➔ **Rad** was the international unit of absorbed dose.

It is used in some old textbooks.

- **1 Gy = 100 Rad** i.e. **1 rad = 1 cGy** or **10 mGy**



kerma

kinetic energy released to mass unit.

- **kerma**, it applies to X- rays , gamma rays & neutrons.
- **Exposure** : was old term to described absorbed dose.
- It was measured by the unit **Rontgen = 10mGy**

Summary – Radiation Quantities & Units

Quantity	Equation	Medium	Type of Radiation	SI unit	Classical unit	Relation
Activity	$A = dN/dt$	Any medium	Any radiation	Bq (dps)	Ci	$1 \text{ Ci} = 3.7 \times 10^{10} \text{ Bq}$
Absorbed dose	$D = dE/dm$	Any medium	Any radiation	Gy (J/kg)	Rad $1 \text{ Rad} = 100 \text{ ergs/g}$	$1 \text{ Gy} = 100 \text{ Rad}$
Equivalent dose	$H = D \times W_R$	Living tissue	Radiation dependent	Sv	rem	$1 \text{ Sv} = 100 \text{ rem}$
Effective Dose	$E = H \times W_T$	Whole body		Sv	rem	$1 \text{ Sv} = 100 \text{ rem}$
Collective effective dose	$S = E_i N_i$			man-Sv	man-rem	
Exposure	$X = dQ/dm$	Air	X, γ	C/kg	Roentgen, R	$1 \text{ R} = 2.58 \times 10^{-4} \text{ C/kg}$

Measurement of X- and gamma ray dose

- It is extremely difficult to measure absorbed dose in solids or liquids directly.
- **In theory**, this can be done by measuring temperature rise,
- **In practice**, temperature change for high absorbed dose of 1 Gy
→ a temperature rise not Significant → it is impractical.

TOOLS OF DOSIMETER

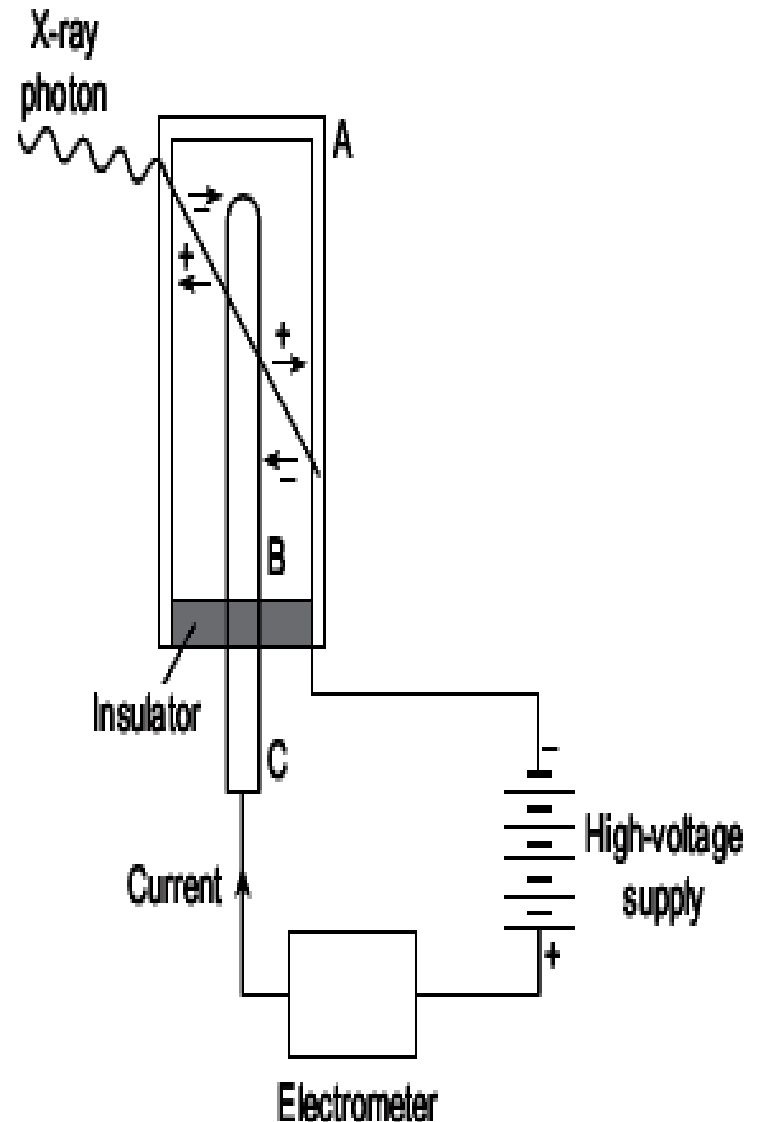
1- Ionization chamber

- **Air kerma**

= measuring the amount of ionization produced by the photon beam in air.

- **Chamber** consists of:

- Plastic outer wall (A)
- Air-filled cavity (B),
- Insulator separating it from
- thin central electrode (C).



- **wall material** : (*air-equivalent material*) matches air of its effective atomic number.
- **Wall thickness** : sufficiently thick → electrons produced outside the chamber will not penetrate the wall & deposit ionization in the cavity;
 - ✓ (**0.2 mm** is sufficient for photoelectrons from 140 keV X-rays)
 - ☒ If the wall is **Too thick**, → **Attenuate** the radiation being measured.

Its use In practice, need to apply *several corrections*,

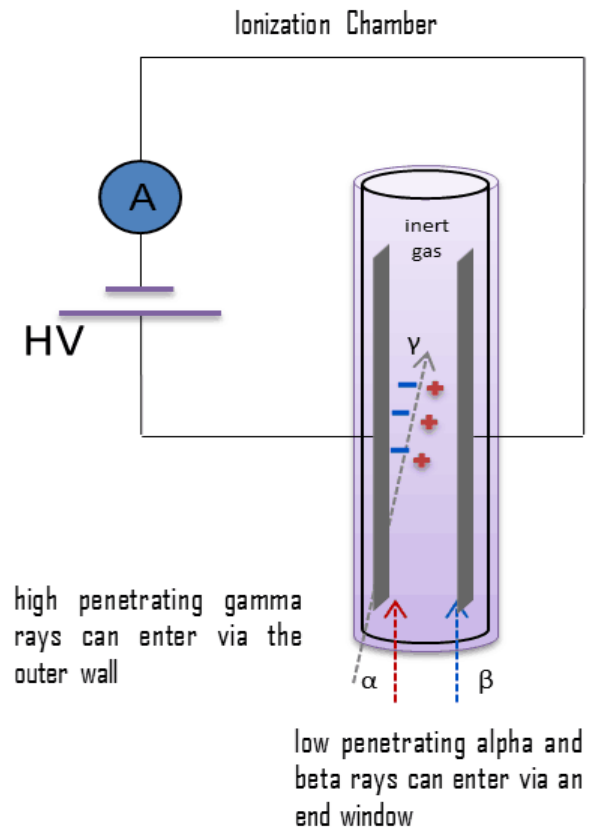
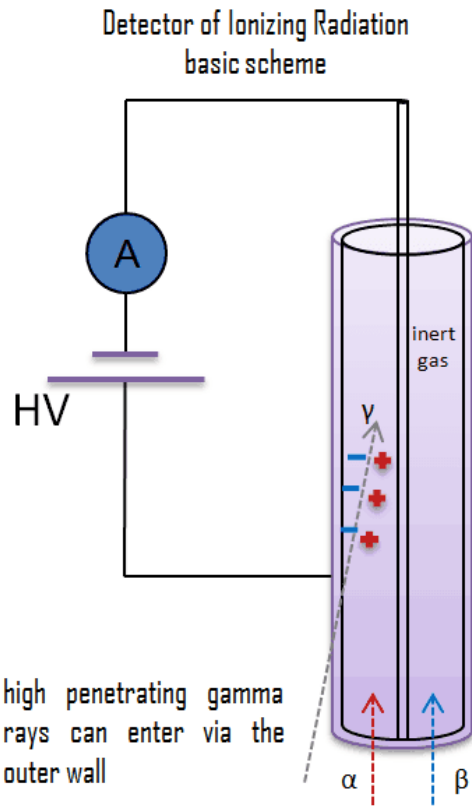
- **Chamber Shape** : commonest → cylindrical (as illustrated here) or consist of parallel electrodes.
- **Chamber volume**: for diagnostic radiology → **10–30 cm³**
for the measurement of scatter radiation → **150 cm³** or bigger.

Why air in ionization chamber ?

- Air is the standard material for dosimetry because:
- it has an effective atomic number (7.6) close to that of tissue (7.4),
→ factor used to convert absorbed dose in air to absorbed dose in tissue can be made **easily** and **accurately**
- it is applicable for measurement over a wide range of X- and gamma ray energies
- large and small doses are easily and accurately measured



Ionization chamber made by Pierre Curie, c 1895-1900



Other dosimeters

- **Lithium fluoride thermo-luminescence dosimeters**

used for both personal and patient dosimetry

- **Photographic effect** in silver bromide, used in film badge

- **Photoconductivity** in silicon diodes to be used in

direct reading electronic personal dosimeters and

dosimeters used for quality assurance.

Radiation quantity and quality

Intensity = the **amount** or **quantity** of radiation

approximately

- proportional to the square of the kV
- proportional to the mA
- inversely proportional to the square of the distance F from a point source.

- these quantities are:
- decreased as the filtration is increased
- greater for a constant potential than a pulsating potential
- ● greater for high rather than low atomic number targets.

Quality

The penetrating power of an X-ray beam.

LUMINESCENCE

The process in which a material absorbs energy from an external source and re-emits that energy in the form of visible light.

- The external energy source may be : chemical, biological and physical sources,
- in radiology we are concerned only with radiation sources for which the term **photoluminescence** may be used.
- Luminescence can be divided into two types:
- **fluorescence**, which is (more or less) the emission of light directly following energy input
- **phosphorescence**, which describes delayed light emission referred to as afterglow.

Fluorescence

- Is the function of intensifying screen of X ray cassette
- It convert X ray to light directly
- X film is more sensitive for light more than X ray
- Thus it reducing dose of x ray & improving image

Sources & Further reading

- <https://petervis.com/electronics%20guides/polonium%20detector/ionisation%20chamber.html>
- https://en.wikipedia.org/wiki/Ionization_chamber
- <https://www.radiation-dosimetry.org/what-is-ionization-chamber-ion-chamber-definition/>



THANK YOU

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